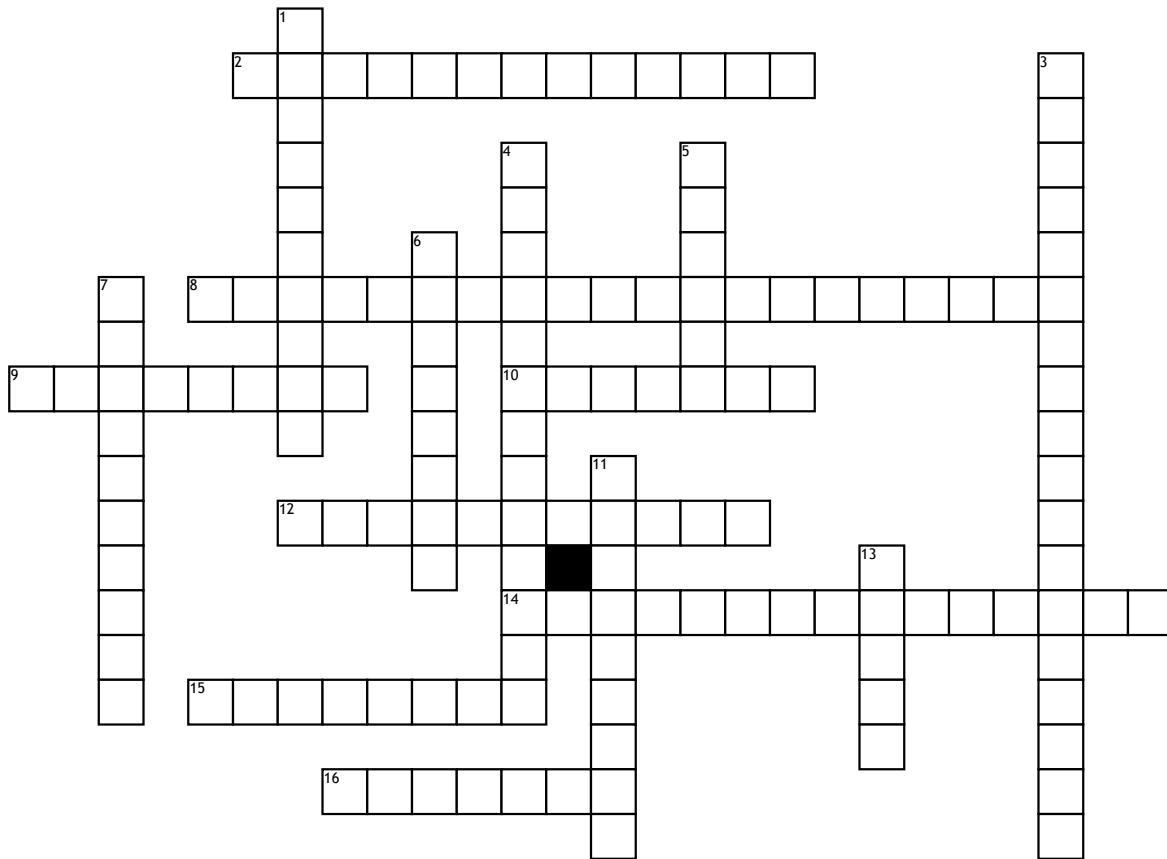


Unit 6AB: Solution Vocabulary



Across

2. a measure of the amount of dissolved substance contained per unit of volume. SEE MORE

8. a mixture with a non-uniform composition.

9. a homogeneous, molecular mixture of two or more substances.

10. a substance that dissolves another to form a solution

12. having a double or triple bond and capable of taking on elements or groups by direct chemical combination without the liberation of other elements or compounds

14. a solution in which the solvent is water.

15. to make a solution of, as by mixing with a liquid; pass into solution

16. susceptible of being dissolved in or as if in a liquid and especially water.

Down

1. the transfer of electricity between two parts of a stationary system, caused by an electrical difference between the parts.

3. a mixture in which the components that make up the mixture are uniformly distributed throughout the mixture.

4. to increase the concentration of (a solution) beyond saturation; saturate abnormally.

5. the substance dissolved in a given solution.

6. containing no permanently dipolar molecules; lacking a dipole.

7. the quality or property of being soluble; relative capability of being dissolved.

11. containing the maximum amount of solute capable of being dissolved under given conditions.

13. Capable of ionizing