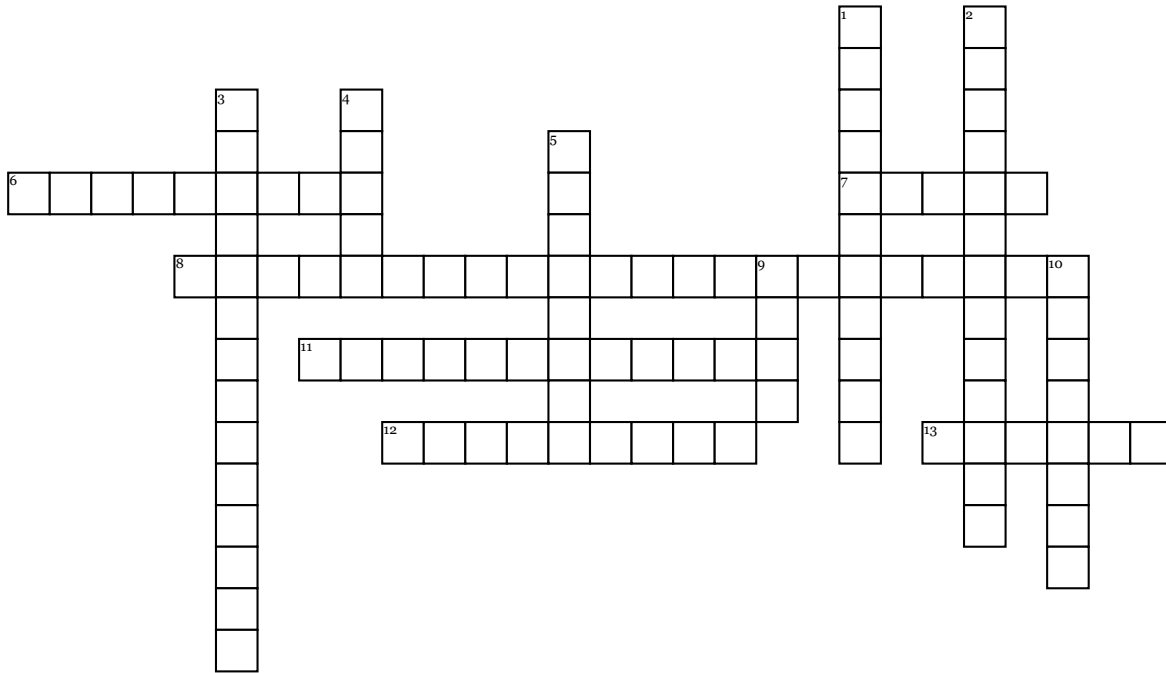


Soluble and Insoluble Salts



Across

6. ions that carry a 3+ charge are

7. salts are _____ compounds

8. soluble salts can be prepared by

11. How is an insoluble salt prepared?

12. ions that do not take part in the reaction are _____ ions

13. direct combination is the preparation of _____ salts

Down

1. ions that can have more than one oxidation states are

2. The double displacement reaction results in?

3. _____ is an example of an insoluble salt

4. solubility rule speaks to

5. compounds that cannot be broken up are?

9. the trivial name for iron oxide is

10. all are soluble