

Name: _____

Date: _____

Period: _____

Revision

B M E T A L H Y D R O X I D E H Y D R O G E N U
E T O I E V I T C A E R E R O M B N H I U Y P G
G O U T E R E L E C T R O N P V E Y U I P C S D
N E T E U N T Y L U L B M 7 O E R A D H I R O D
A V H Y O O U Y S N C P V D O P L 7 H I H G S U
H I S T 7 I V V Y R C R N 7 G 7 S A L T S V V D
C T E I C T A R Y E G M Y E D I X O L A T E M D
R C I V H C Y S I A S D I S P L A C E M E N T Y
U A R I E A P X N C V T M A D G E P Y H L L X L
O E E T V R U I M T S E E I R Y A 7 G G I T M H
L R S C I T S T C I D X E A M H H P G O A V O T
O S Y A T T G L X V P 7 B I S 7 V U N B 7 N P T
C S T E C A U H C E Y X B X H P O O I B M X P A
O E I R A H G G M X N M T L I A M R L S P 7 N M
V L V Y E H T V H A X A M U U P U G B E A C U N
M O I S R S B C C U C U E L H X 7 H B L N X C I
N V T E I X V L H U Y D H G I A I R U B T I L G
I G C G C L D G N C P U D X Y I P D B I 7 H E E
A G A V A O C H 7 A 7 O P L I T I S E X 7 7 U I
M C E I N O R E A C T I O N P D R L S Y N G S E
7 L R Y V E U V D I B E 7 H U P S N E G O L A H
A B P V G U L M E T A L T 7 O D I I 7 I C G T N
G I B S 7 T N I L I I I 7 L R Y S P C U 7 E Y B
B A R Y A X P H Y R I A O H G N I A U L O I S L

metal hydroxide + hydrogen
colour change
displacement
unreactive
bubbling
nucleus
metal

reactivity series
less-reactive
no reaction
attraction
reactive
group 7
salts

outer electron
more-reactive
metal oxide
reactivity
halogens
group 1