

Name: _____

Date: _____

Redox SL

F S U O P G T T S A D I C J R S N
E A N X E Y S P O N T A N E O U S
L L C I E L E C T R O D E A O V W
E T A D W E G B M O L T E N F U A
C B T A R E D U C T I O N O O G F
T R H T E O E P T E O G C D T R G
R I O I X M T L G L V A F E Q A A
O D D O T L V A A E A U Z U P P L
L G E N E G W T I C I G L H A H V
Y E P C R N Q I N T Z C R C H I A
T I B Y N C U N E R J G V X H T N
I Z B U A S E U R O Q H H S G E I
C P W F L R V M T L X B V M Y Y C
C T B L W A O A H Y I R W E J N C
E P C X I C C A Y T H A I A B D E
L I L U R W D C T E H N S K Q P L
L F Q J E O V O L T A I C C E L L

Electrolytic cell
spontaneous
oxidation
cathode
CRC

external wire
Electrolyte
electrode
molten
AOA

Galvanic cell
salt bridge
platinum
inert

voltaic cell
reduction
graphite
anode