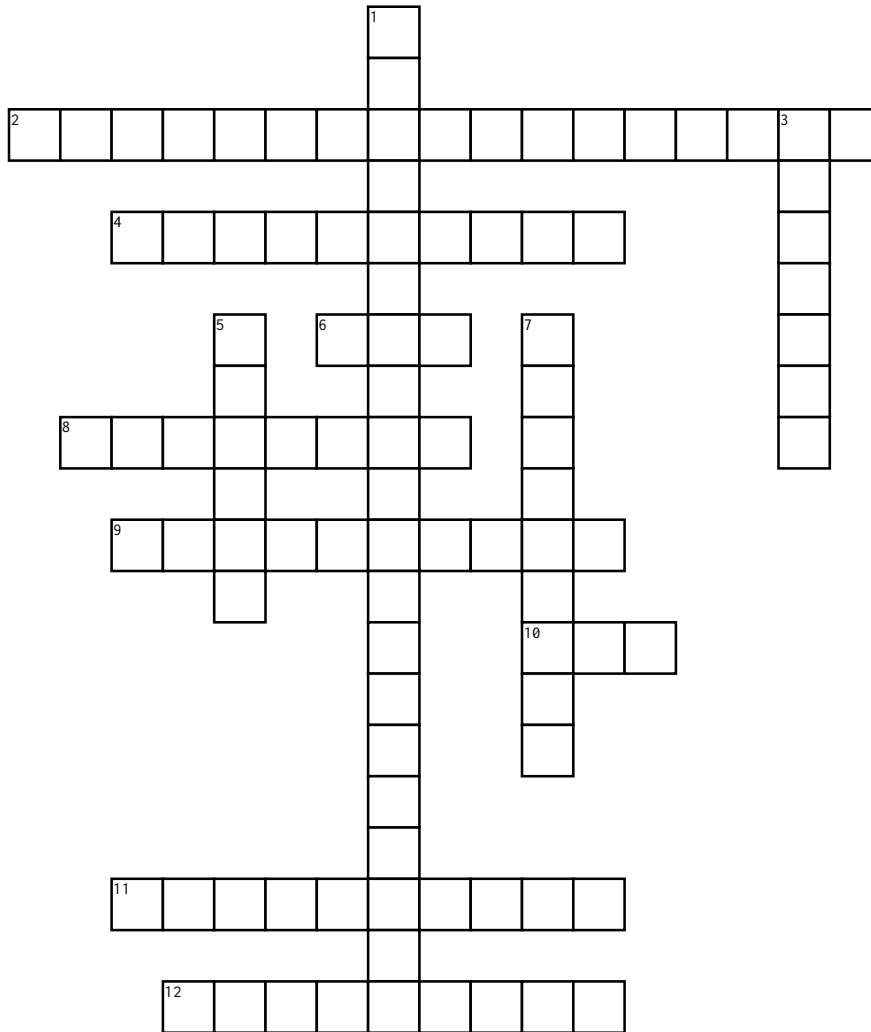


Periodic Table and Compound Review



Across

2. What is the order for a Covalent Compound?
 4. NO₃⁻ is an example of a ____ ion
 6. What Common Polyatomic Ion is nitrite?
 8. atoms of the same element with different numbers of neutrons
 9. An ion with 2 or more atoms are called ____ ions

Word Bank

Isotopes
 polyatomic
 NO₂⁻
 transition

10. In a Single Charged Binary Compound, nonmetals change to ____.
 11. The elements in the middle of the periodic table are called _____ metals
 12. covalent compounds are made of

electronegativity
 Ternary
 polyatomic
 metals

Down

1. When naming a covalent compound, what is the prefix for N₂O₄?
 3. If a compound has more than 2 elements its called a _____ compound.
 5. When naming Single Charged Binary Compounds, _____ come first.
 7. Chemical bonds form in order to increase..

nonmetals
 -ide
 stability
 dinitrogen tetroxide