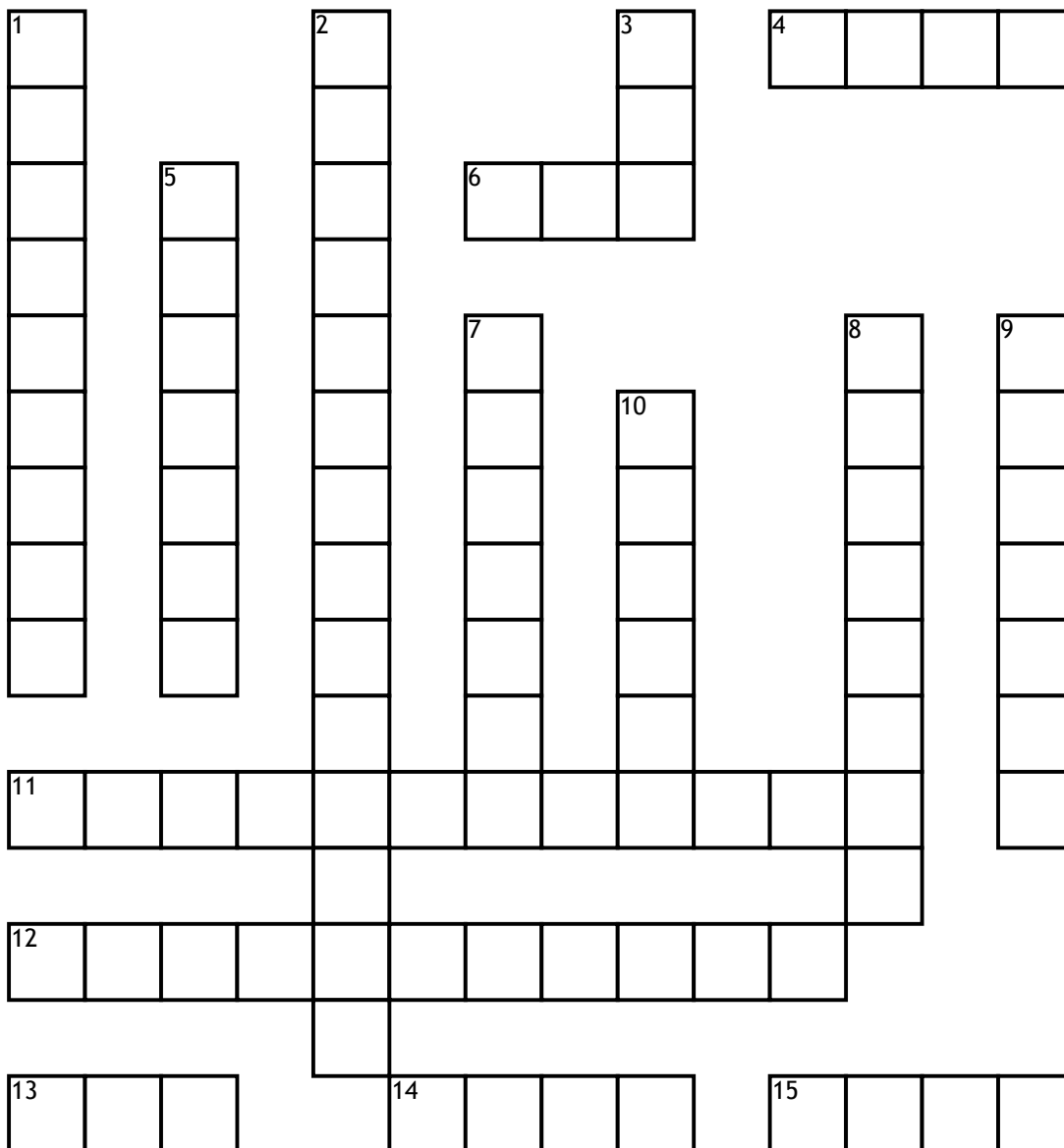


Neutralization Reactions



Across

- 4. A pH below 7
- 6. The compound that contains oxygen ends in this
- 11. An Ion that does not react in the equation
- 12. The product of $H_3O + CH_3CO_2$
- 13. A compound ending in this is mono-atomic
- 14. any compound created by the reaction between an acid and base

15. A pH above 7 (MAX 14)

Down

- 1. What reactant would give you $H_2O + NaNO_3$?
- 2. what happens when you add a strong acid to a strong base?
- 3. A compound name ending in this has one less oxygen than it would if it ended in -ate
- 5. What would the product of $H + HCO_3$ be?

7. The final part of the equation

8. The first part of the equation

9. A solution with a solvent of water

10. The reactant of $HBr + H_2O$?