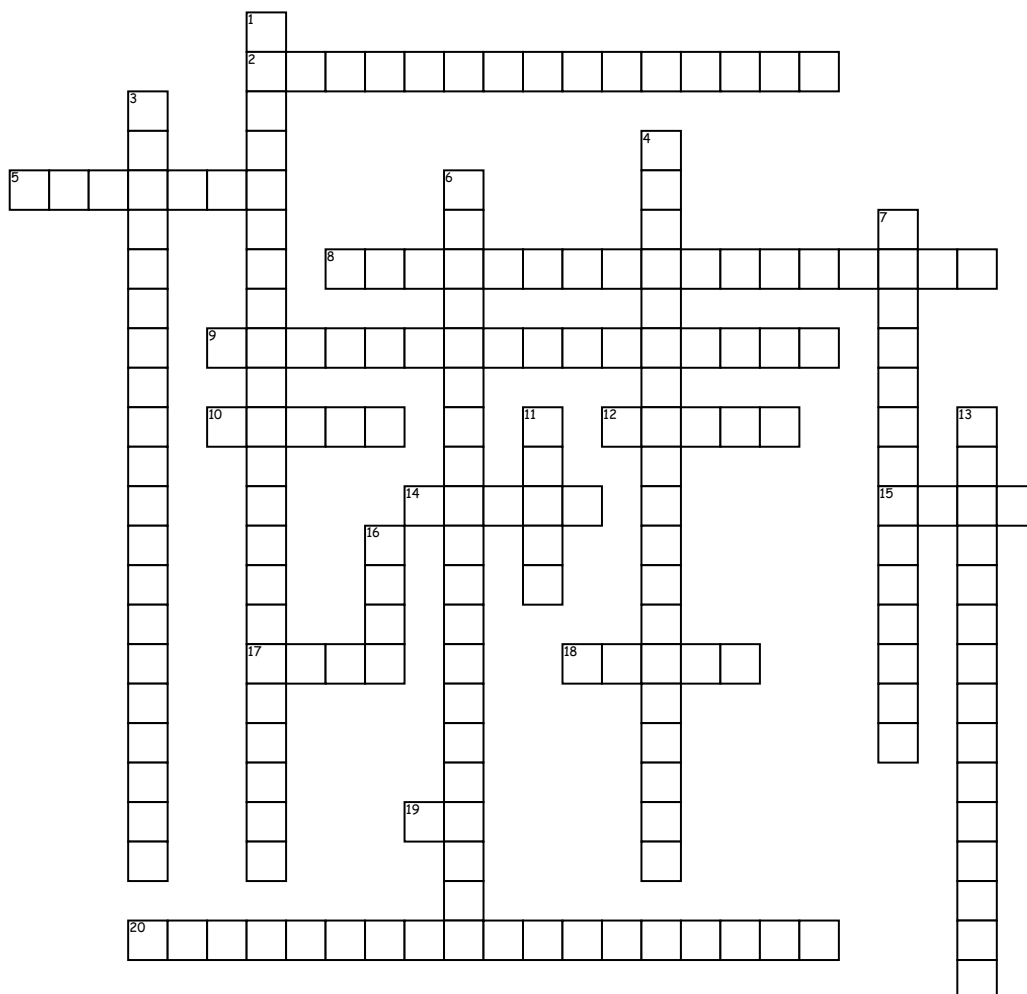


Name: _____

Naming Ionic and Covalent Compounds



Across

2. Al_4C_3
 5. NH_3
 8. XeF_6
 9. $Be(NO_3)_2$
 10. Nickel (ii) Sulfate
 12. Titanium (iii) Bromide
 14. Iron (iii) Sulfide

15. Sulfur Hexachloride
 17. Tin (iv) Oxide
 18. tetraphosphorus triselenide
 19. Carbon Monoxide
 20. N_2O_3

Down

1. $Mn_3(PO_4)_2$

3. PH_3
 4. $Co_3(PO_4)_2$
 6. BrF_5
 7. Li_2S
 11. Chromium (ii) Sulfate
 13. CaF_2
 16. Disulfur Difluoride

Word Bank

CO	Bromine Pentafluoride	SnO ₂	S ₂ F ₂
Calcium Fluoride	NiSO ₄	Cobalt (ii) Phosphate	Ammonia
Xenon Hexafluoride	SnI ₆	P ₄ Se ₃	Dinitrogen Trioxide
Beryllium Nitride	Fe ₂ S ₃	TiBr ₃	Aluminum Carbide
Manganese (ii) Phosphate	Phosphorus Trihydride	CrSO ₄	Lithium Sulfide