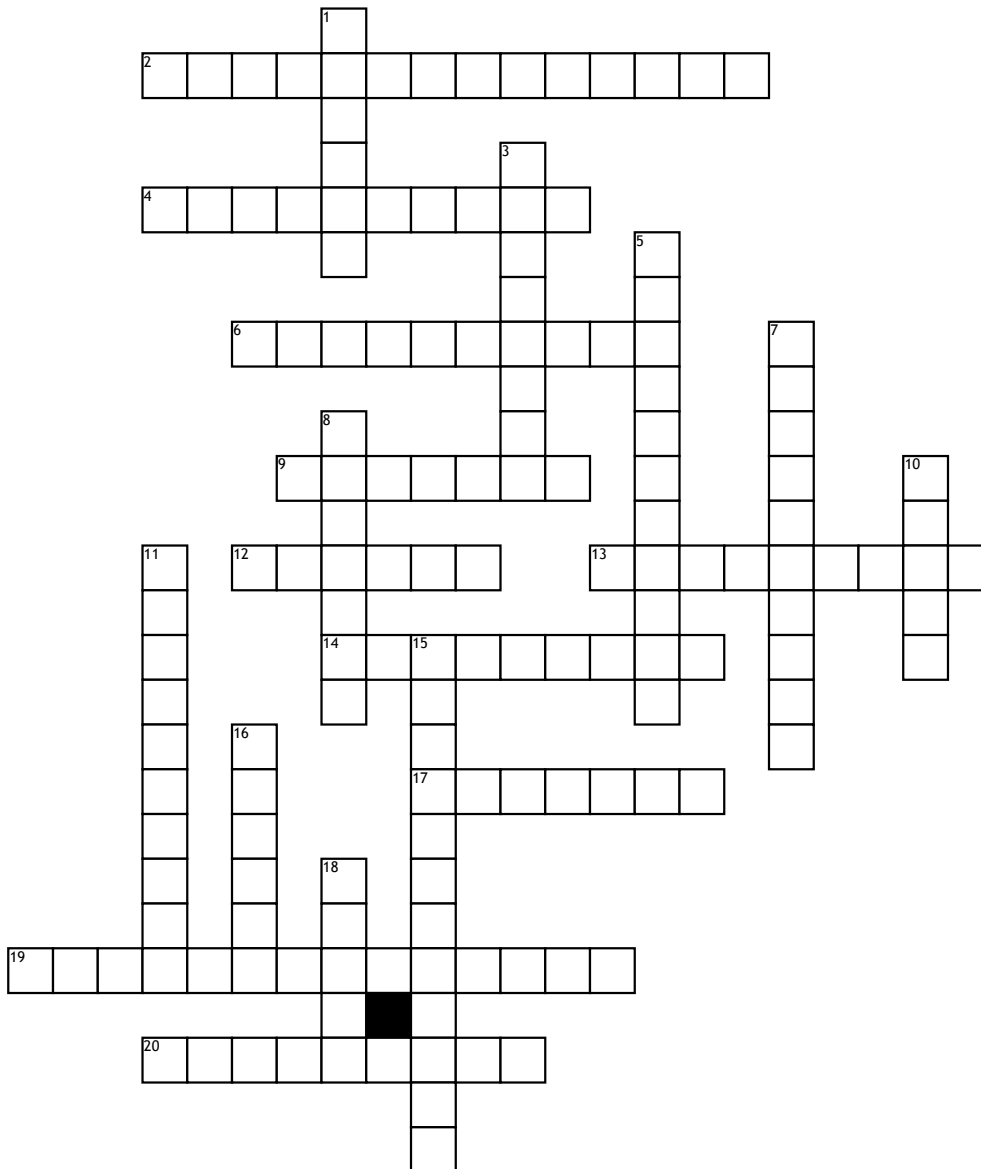


Mixtures and Solutions



Across

- 2. Solubility can be used to identify a substance because it is a ____ property.
- 4. Mixtures in which particles can be seen and separated by settling or filtration
- 6. Suspensions do not have the same ____ throughout.
- 9. Part of a solution present in the largest amount
- 12. Type of solution with only a little solute
- 13. A solution contains solute ____ too small to see.
- 14. As temperature increases, solubility ____ .

17. The ____ in a car radiator is a solution of water and antifreeze.

- 19. Crystals will form in a ____ solution when excess solute is added.
- 20. Type of solution where no more solute will dissolve in it

Down

- 1. A colloid contains ____ particles than a solution
- 3. A homogeneous mixture that contains a solvent and at least one solute
- 5. Type of solution to which you can continue to add more solute
- 7. Measure of how much solute can be dissolved in a solvent at a given temperature

8. A mixture containing small, undissolved particles that do not settle out

- 10. ____ water on the surface of a lake can be frozen, but the ocean cannot freeze.
- 11. An example of a colloid
- 15. Solution with a lot of solute dissolved in the solvent
- 16. Solutions may be any combination of gases, liquids, or ____ .
- 18. ____ is the solvent in blood, saliva, and tears.