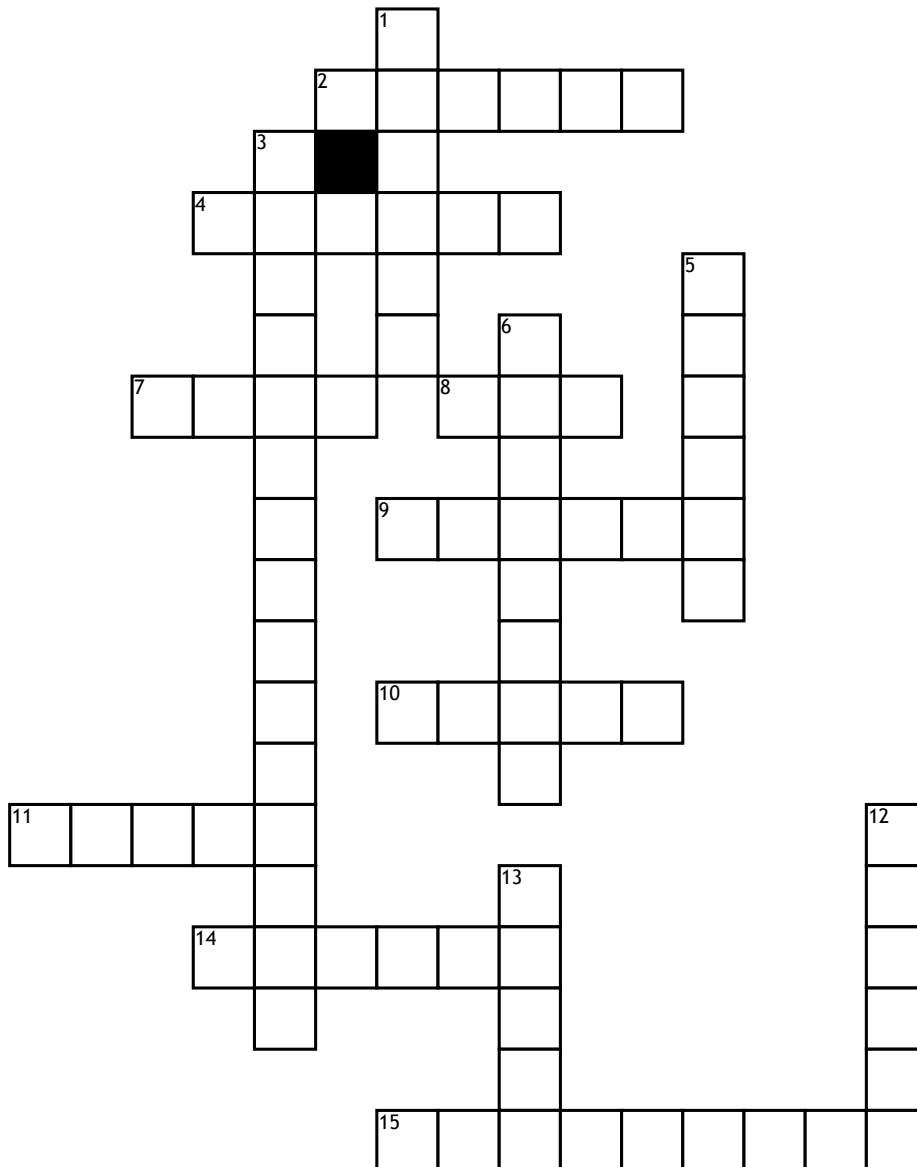


Bonding Review



Across

2. The element that has 6 energy levels but has 2 valence electrons
4. A group of elements that act the same way
7. An attraction towards another element is called a _____
8. There are 7 total energy levels, each one represented on the periodic table by a _____

9. An ion is any element that has a _____?

10. Type of bond where electrons are given/taken
11. The number of valence electrons you must have to obey the octet rule
14. To determine an elements valence electrons you can look at the _____ on the periodic table
15. Elements on the right side of the periodic table

Down

1. A positive ion
3. Electron on the outermost energy level
5. The element that has 2 energy levels but has 6 valence electrons
6. Type of bond where electrons are shared
12. The elements on the left side of the periodic table
13. A negative ion