

Name: _____

Atomic Theory

E O E P H R E B M U N S S A M J J T H O M S O N
R G L L R N O T L A D N H O J B R O M I N E A E
N N E K X E E K D E L B A T C I D O R E P N L K
E X C G Z M I K C I W D A H C S E M A J A P C D
S O T L C W S O E C N A L A V G K T P H X K L Q
T I R M B A T D D J F A N O B L E G A S E S B H
R K O O L R E B M U N C I M O T A A C Q A B J R
U R N G A M F T N E M E L E K B H E K P O G L C
T Q S N O I T A R U G I F N O C N O R T C E L E
H K H O R P W R S I O N I C B O N D V Q Y R T F
E S E P D M S N W N D F L Q S V A T O M O S B M
R G L R L U E P K C H E M I C A L F O R M U L A
F J L K E G R R P R G S K I I F A I M E G U R H
O P C L O O B M C N F D U L R N O I S N X D H Y
R D C L T D E G P U H R S T A B Q U R K Y L O N
D U A O F P M U H C R G C M I N M P U O R G B E
N H N J O S Y H D O N Y W N R R T E J W K L S U
S L A T E M N O I T I S N A R T C H M Q N Q L T
H Q O G M E N D E L E E V J K L Y O A Q Q D I R
S S U I E L E C T R O S T A T I C M M N G B E O
I W C X U Y R C K T N O R T C E L E E E I Z N N
T S H T R A E E N I L A K L A A J Z E V D D J X
H X L H D S L A T E M I L A K L A D B D P F E W
U C M O T A H B X D N U O P M O C R K P B P X S

electronconfiguration
alkaline earths
electrons
shell
mass number
Neils Bohr
compound
isotope
neutron
atomos

transition metals
James Chadwick
atomic number
John Dalton
JJ Thomson
halogens
bromine
nucleus
group

Ernest Rutherford
electrostatic
alkalimetals
ionic bond
Democritus
electron
mercury
preiod
atom

chemical formula
Perodic Table
lanthanides
noblegases
Mendeleev
element
valance
proton
ion