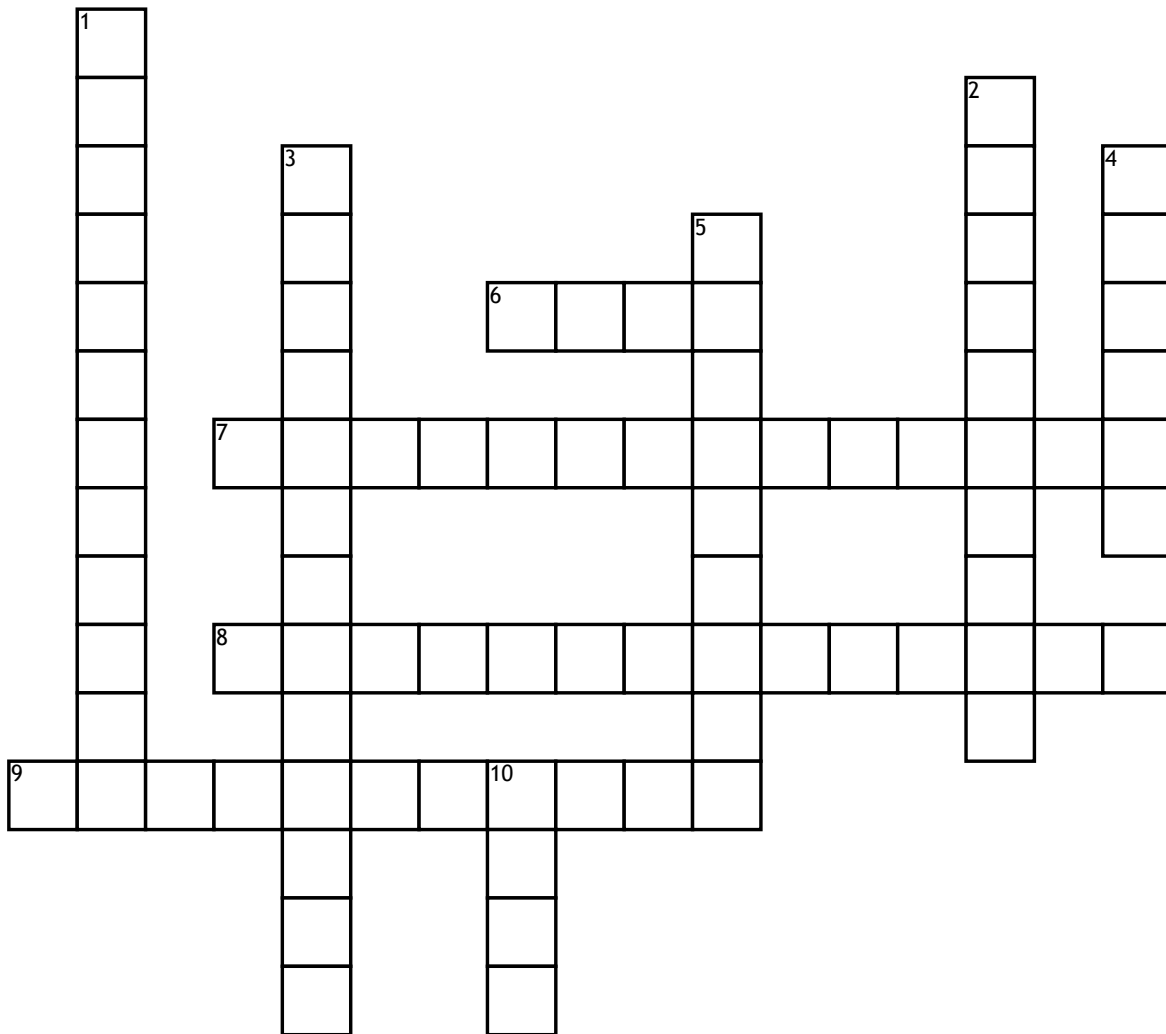


Acids-bases



Across

6. A substance that can accept hydrogen ions
7. The reaction of an acid with a base to produce a salt and water
8. Bronsted-Lowry definition of base
9. A solution where more solute could be dissolved

Down

1. The positive ion formed when a water molecule gains a hydrogen ion.

2. The biggest amount of a substance which can be dissolved in a 100 g of water
3. The amount of substance in the amount of solution
4. An acid that is completely ionized in aqueous solution
5. A substance that acts as an electron acceptor
10. A substance that can donate hydrogen ions